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Spectroscopy of Rydberg atoms in a Black-body radiation field: Relativistic theory of excitation and ionization

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Abstract. The combined relativistic energy approach and many-body perturbation theory with zeroth model potential approximation are used for computing Blackbody radiation ionization characteristics of the Rydberg atoms, in particular, the sodium in states with n=17,18,40-70. The calculated ionization rate values are compared with available theoretical and experimental data.

1. Introduction

A great progress in experimental laser physics and appearance of the so called tunable lasers allows one to access the highly excited Rydberg states of atoms. In fact this is a beginning of a new epoch in the atomic physics with external electromagnetic field. It has stimulated a great number of papers on the ac and dc Stark effect [1-12]. Conversely, the experiments with Rydberg atoms had very soon resulted in the discovery of an important ionization mechanism, provided by unique features of the Rydberg atoms. Relatively new topic of the modern theory is connected with consistent treating the Rydberg atoms in a field of the Blackbody radiation (BBR). It should be noted that the BBR is one of the essential factors affecting the Rydberg states in atoms [1]. The account for the ac Stark shift, fast redistribution of the levels' population and photoionization provided by the environmental BBR became of a great importance for successfully handling atoms in their Rydberg states.

The most popular theoretical approaches to computing ionization parameters of the Rydberg atom in the BBR are based on the different versions of the model potential (MP) method, quasiclassical models. We mention a simple approximation for the rate of thermal ionization of Rydberg atoms, based on the results of our systematic calculations in the Simons-Fues MP [1]. In fact, using the MP approach is very close to the quantum defect method and other semi-empirical methods, which were also widely used in the past few years for calculating atom–field interaction amplitudes in the lowest orders of the perturbation theory. The significant advantage of the Simons-Fues MP method in comparison with other models is the possibility of presenting analytically (in terms of the hypergeometric functions) the characteristics for arbitrarily high orders, related to both bound–bound and bound–free transitions. Naturally, the standard methods of the theoretical atomic physics, including the Hartree-Fock and Dirac-Fock approximations, should be used in order to determine the thermal ionization characteristics of neutral and Rydberg atoms [2]. Note that the correct treatment of the heavy Rydberg atoms parameters in an external electromagnetic field requires using strictly relativistic models. Here we apply an energy approach [11-15] and relativistic perturbation theory (PT) with the MP zeroth approximation [16-20] to computing the thermal BBR ionization characteristics of the Rydberg atoms, in particular, the sodium.

2. Ionization of the Rydberg atoms in the Blackbody radiation

The qualitative picture of the BBR Rydberg atoms ionization is in principle easily understandable. Even for temperatures on the order of $T=10^4$ K, the frequency of a greater part of the BBR photons ω does not exceed 0.1 a.u. One could use a single- electron approximation to calculate the ionization cross section σ_{nl} (ω). The latter appears as a product with the Planck's distribution for the thermal photon number density:

$$\rho(w,T) = \frac{\omega^2}{\pi^2 c^3 [\exp(\omega/kT) - 1]},\tag{1}$$

where $k=1.3807 \times 10^{-23}$ J·K⁻¹ is the Boltzmann constant, $c = 2.9979 \cdot 10^8$ m·s⁻¹ is the speed of light. Ionization rate of a bound state *nl* results in the integral over the Blackbody radiation frequencies:

$$P_{nl}(T) = c \int_{|E_{nl}|}^{\infty} \sigma_{nl}(\omega) \rho(\omega, T) d\omega.$$
⁽²⁾

The ionization cross-section from a bound state with a principal quantum number n and orbital quantum number l by photons with frequency ω is as follows:

$$\sigma_{nl}(\omega) = \frac{4\pi^2 \omega}{3c(2l+1)} [lM_{nl\to El-1}^2 + (l+1)M_{nl\to El+1}^2], \qquad (3)$$

where the radial matrix element of the ionization transition from the bound state with the radial wavefunction $R_{nl}(r)$ to ith continuum state with wavefunction $R_{El}(r)$ is normalized to the delta distribution of energy. The corresponding radial matrix element can be written as:

$$M_{nl\to El'} = \int_{0}^{\infty} R_{El'}(r) r^{3} R_{nl}(r) dr.$$
(4)

We apply a generalized energy approach [11-15] and relativistic perturbation theory with the MP zeroth approximation [16-20] to computing the Rydberg atoms ionization parameters. In relativistic theory radiation decay probability (ionization cross-section etc) is connected with the imaginary part of electron energy shift. The total energy shift of the state is usually presented in the form: $\Delta E = \text{Re}\Delta E + \text{i} \Gamma/2$, where Γ is interpreted as the level width, and a decay probability P = Γ . The imaginary part of electron energy shift is determined as [13, 14]:

$$\operatorname{Im} \Delta E = -\frac{1}{4\pi} \sum_{\substack{\alpha > n > f \\ [\alpha < n \le f]}} V_{\alpha n \alpha n}^{|\omega_{\alpha n}|}, \qquad (5)$$

where $\omega_{\alpha n}$ is a frequency of the α -n radiation, ($\alpha > n > f$) for electron and ($\alpha < n < f$) for vacancy. The matrix element *V* is determined as follows:

$$V_{ijkl}^{[\omega]} = \iint dr_1 dr_2 \Psi_i^*(r_1) \Psi_j^*(r_2) \frac{\sin[\omega] r_{12}}{r_{12}} (1 - \alpha_1 \alpha_2) \Psi_k^*(r_2) \Psi_l^*(r_1)$$
(6)

Detailed description of the matrix elements and computational procedures are presented in Refs. [12,13,15]. The relativistic wave functions are calculated by solution of the Dirac equation with the potential, which includes the "outer electron - ionic core" potential (in the Miller-Green form [21]) and exchange-polarization potential [20]. All calculations utilize the Superatom-ISAN (version 93) code.

3. Results

In Table 1 we present results of the ionization rate calculation for the Rydberg sodium atom in the states (17,18D, 18P) at temperatures of 300 K and 500 K: Th5 – our (relativistic MP theory) data, E1 – experimental data by Kleppner etal and Burkhardt etal [4], Th1- theory (nonrelativistic Simons-Fues

MP) by Glukhov-Ovsyannikova [9], Th2- theory of Lehman [8], Th3- quasiclassical model by Dyachkov-Pankratov [10] and Th4- theory by Beterov et al. [1]. Overall, there is physically reasonable agreement between the theoretical and experimental data. Obviously, the accuracy of the theoretical data is provided by a correctness of the corresponding relativistic wave functions and accounting for the exchange-correlation effects. In Table 2 we list our results of ionization rate (s⁻¹) for sodium Rydberg states (with n=40-70) induced by BBR radiation (T = 300 K).

Table 1. Theoretical and experimental values of the ionization rate (10³ s⁻¹) of sodium Rydberg states: E1- Kleppner et al Burkhardt etal; Th1 – theory by Glukhov-Ovsyannikova; Th2- theory by Lehman, Th3- theory by Dyachkov-Pankratov; Th4- theory by Beterov et al; Th5 – this work.

Т	nL	E1	Th1	Th2	Th3	Th4	Th5
300	17D	10^{3}	$1.08 \cdot 10^{3}$	$0.95 \cdot 10^3$	$0.9 \cdot 10^3$	$1.147 \cdot 10^3$	$1.02 \cdot 10^{3}$
500	18P	-	$4.18 \cdot 10^3$	-	-	-	$5.54 \cdot 10^3$
500	18D	-	$4.07 \cdot 10^3$	-	-	-	$5.46 \cdot 10^3$

Table 2. Ionization rate (s⁻¹) for the sodium Rydberg states (with n = 40-70), induced by BBR radiation (T = 300 K; our data).

Atom	40	50	60	70
Na S	142	106	61.4	29.5
Na P	804	576	311	141
Na D	707	496	268	122

References

- [1] Beterov I I, Tretyakov D V, Ryabtsev I I, Entin V M, Ekers A and Bezuglov N N 2009 *New J*. *Phys.* **11** 013052
- [2] Safronova U I and Safronova M S 2009 Phys. Rev. A 79 022512
- [3] Gallagher T F 1979 Phys. Rev. Lett. 42 835
- [4] Killian T, Kulin S, Bergeson S, Orozco L, Orzel C and Rolston S 1999 Phys. Rev. Lett. 83 4776
- [5] Li W, Noel M, Robinson M, Tanner P and Gallagher T 2004 Phys. Rev. A. 70 042713
- [6] Spencer W P, Vaidyanathan A, Kleppner D and Ducas T 1982 Phys. Rev. A. 26 1490
- [7] Glushkov A V, Ambrosov S, Ignatenko A and Korchevsky D 2004 Int. J. Quant. Chem. 99 936
- [8] Lehman G W 1983 J. Phys. B: At. Mol. Phys. 16 2145
- [9] Glukhov I and Ovsiannikov V 2009 J. Phys. B: At. Mol. Phys. 42 075001
- [10] D'yachkov L G and Pankratov P M 1994 J. Phys. B: At. Mol. Opt. Phys. 27 461
- [11] Ivanova E P and Ivanov L N 1979 Atom. Data. Nucl. Data. Tabl. 24 95
- [12] Glushkov A V 2012 J. of Phys.: Conf. Ser. 397 012011
- [13] Ivanov L N, Ivanova E P and Aglitsky E V 1988 Phys. Rep. 166 315
- [14] Glushkov A V and Ivanov L N 1992 Phys. Lett. A 170 33
- [15] Glushkov A V 2012 Advances in the Theory of Quantum Systems in Chemistry and Physics. Ser. Progress in Theoretical Chemistry and Physics, vol. 28, ed. by K Nishikawa, J Maruani, E Brandas, G Delgado-Barrio, P Piecuch (Berlin: Springer) p 131
- [16] Glushkov A V, Malinovskaya S V, Prepelitsa G P and Ignatenko V M 2005 J. of Phys.: Conf. Ser. 11 199
- [17] Malinovskaya S V, Glushkov A V, Khetselius O Yu, Svinarenko A A, Mischenko E V and Florko T A 2009 Int. J. Quant. Chem. 109 3325
- [18] Glushkov A V, Khetselius O, Loboda A and Svinarenko A 2009 Phys. Scr. T153 014029
- [19] Khetselius O 2012 J. of Phys.: Conf. Ser. 397 012012; 2009 Int. J. Quant. Chem. 109 3330
- [20] Malinovskaya S V, Glushkov A V, Khetselius O Yu, Loboda A V, Lopatkin Yu M, Svinarenko A A, Nikola L V and Perelygina T B 2011 Int. J. Quant. Chem. 111 288
- [21] Miller K J and Green A E S 1974 J. Chem. Phys. 60 2617